

Nanoparticle Formation in Pores Spaces

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Pore spaces and weathering

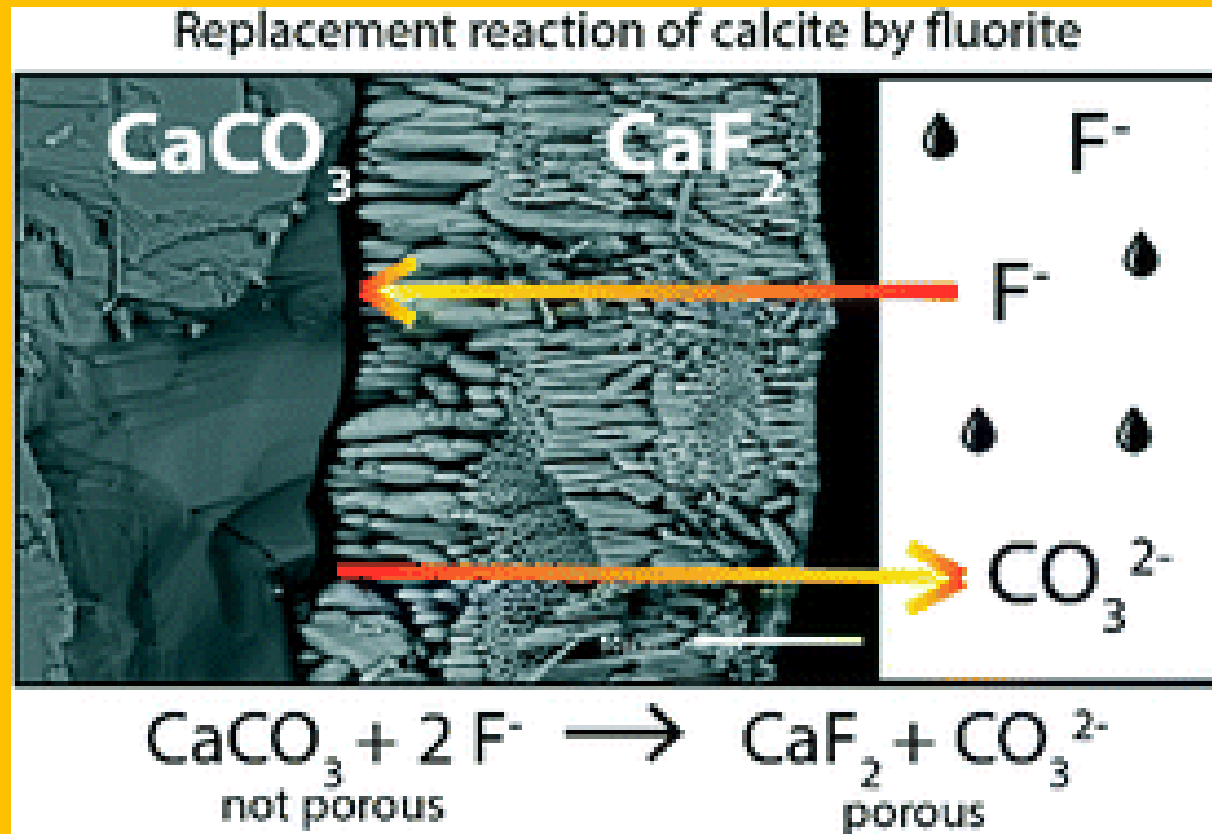
Nanoparticle formation in pore spaces
under hydrothermal conditions

Nanoparticle formation in pore spaces
at low Temperature

Conclusions

Weathering is a result of the re-equilibration of minerals due to changes in e.g. fluid composition, pH, Eh, T, P...

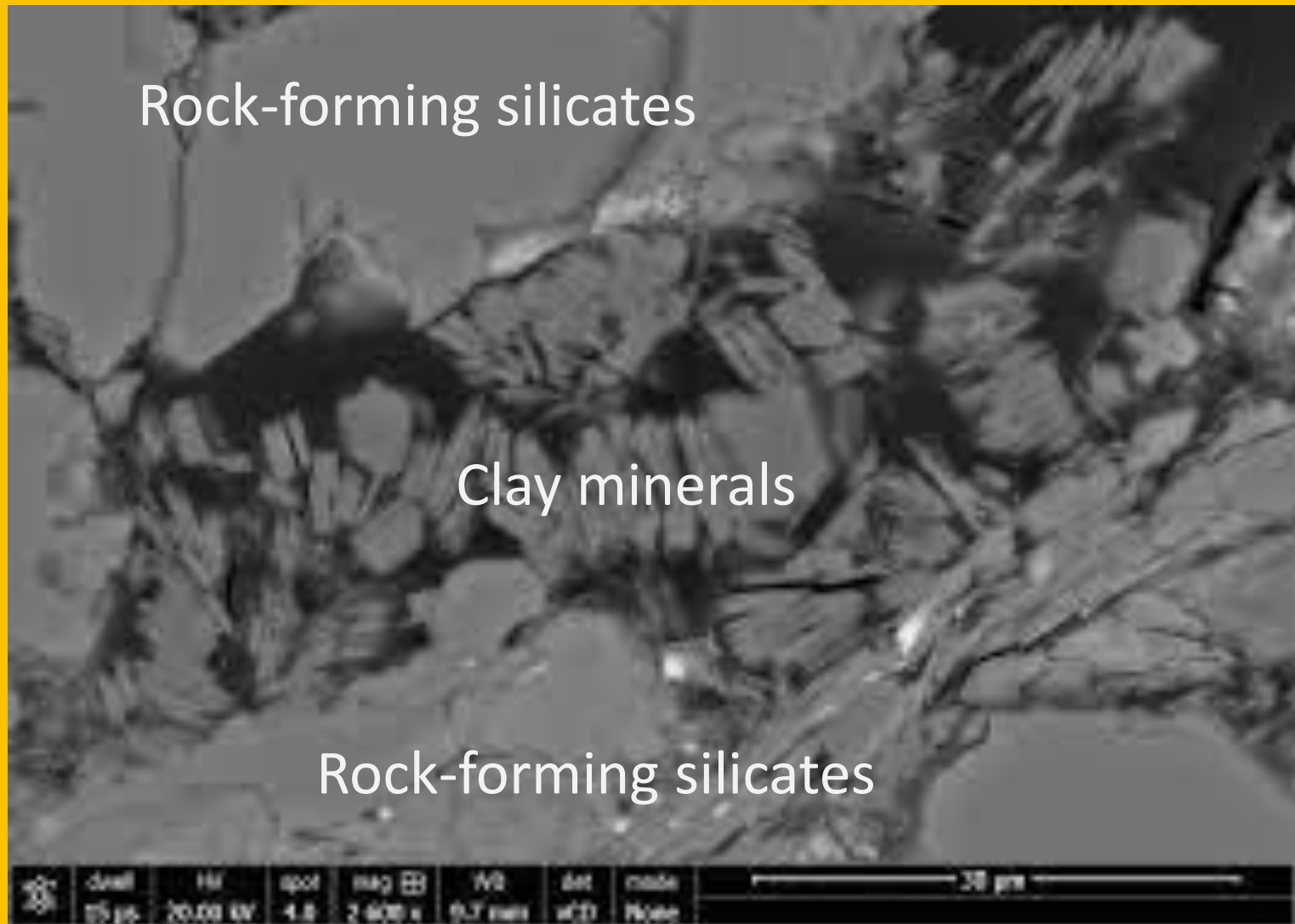
It involves often dissolution-re-precipitation processes with the replacement of a non-porous mother-phase by a porous daughter phase(s)



Pedrosa et al. (2016) Porosity generated during the fluid-mediated replacement of calcite by fluorite.

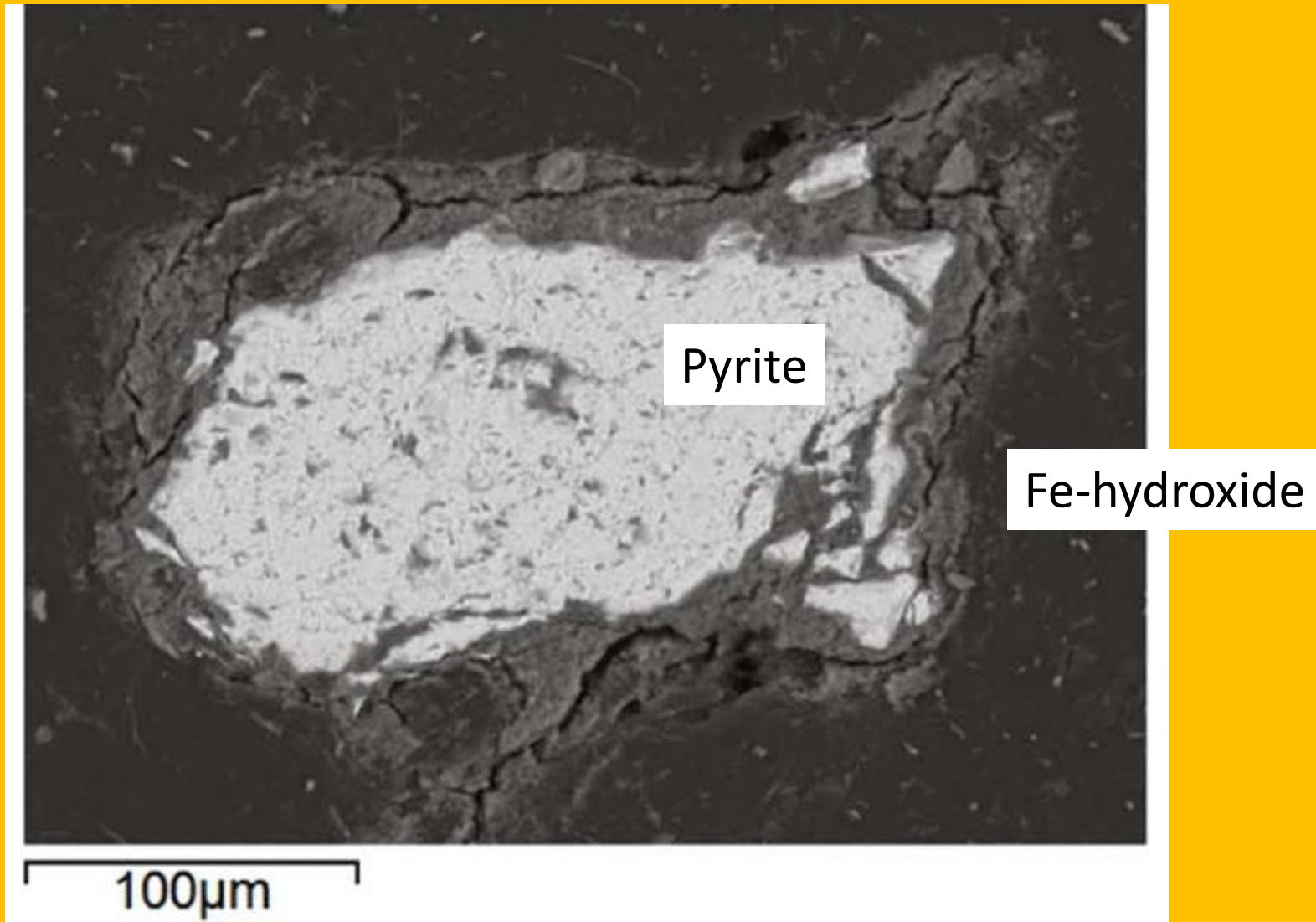
***CrystEngComm*, 2016, **18**, 6867-6874**

Porous Clay mineral surface coating on rock-forming silicates

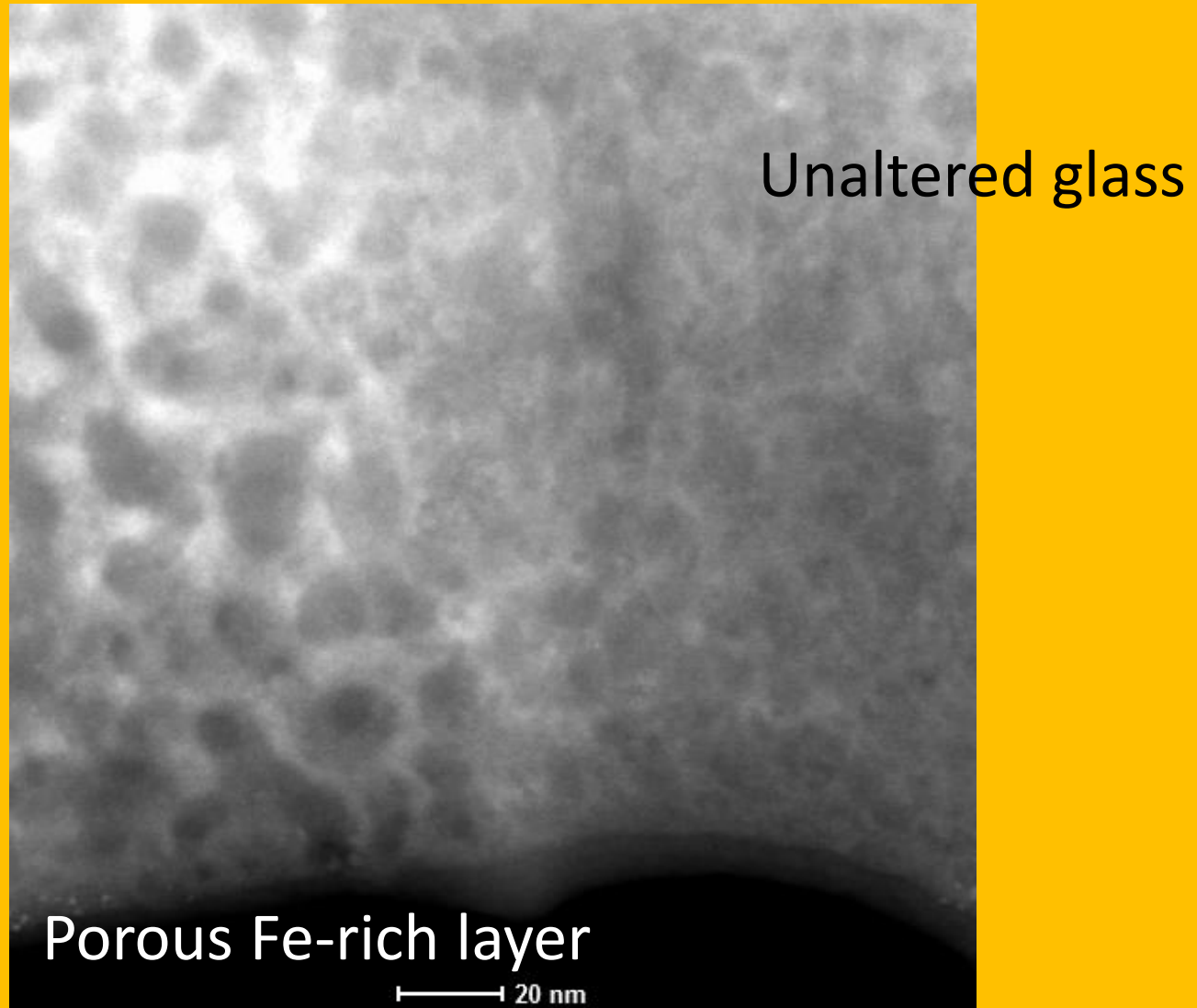


https://files.isgs.illinois.edu/sites/default/files/files/Giannetta_NCGSA_Presentation.pdf

Porous Fe-hydroxide coating on pyrite



Alteration of volcanic glass



Nanoparticle formation during mineral replacement reactions under hydrothermal conditions

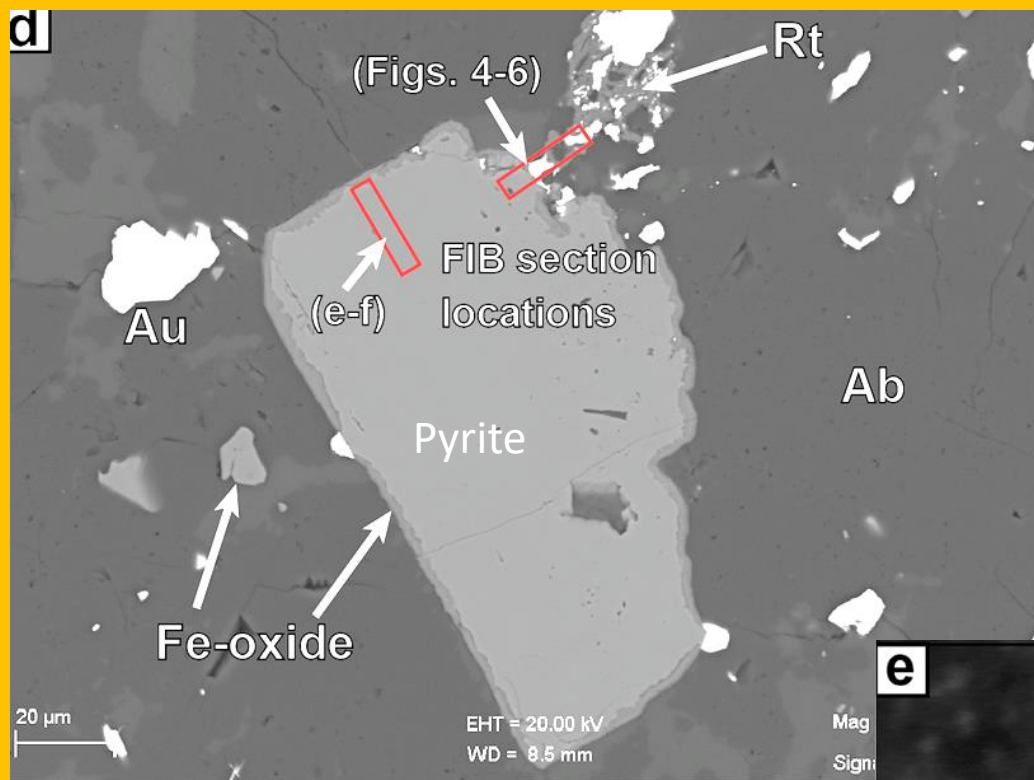
Two examples:

Gold nanoparticles and high-grade gold ore deposits

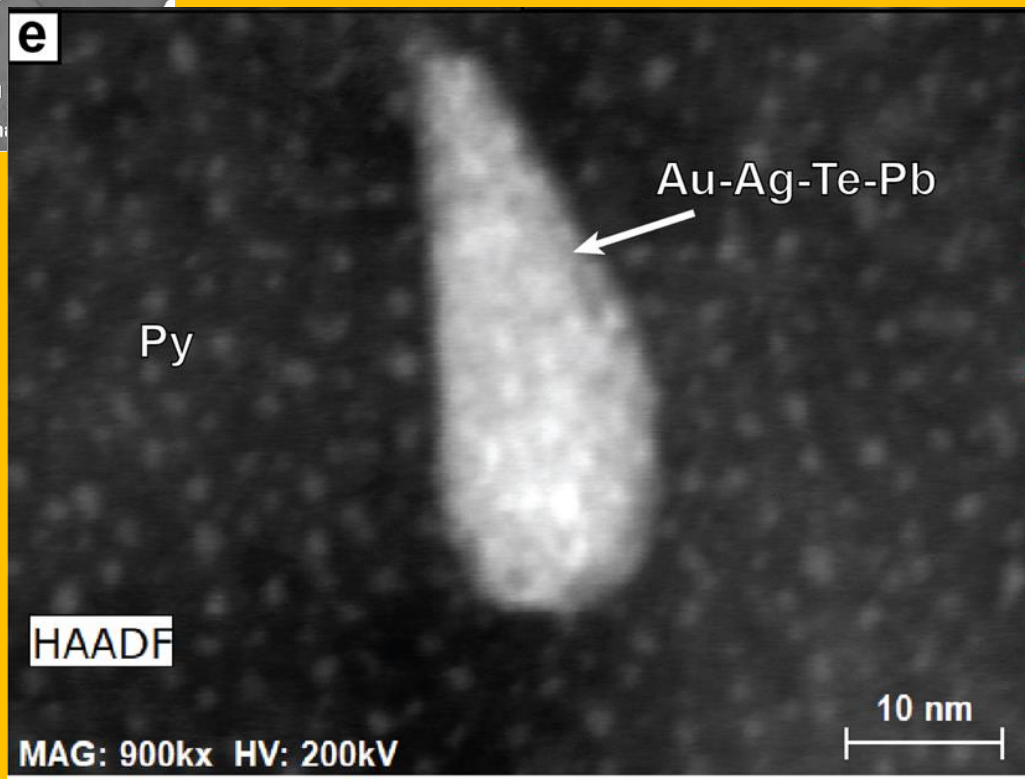
Chromite nanoparticles during greenschist metamorphism

High grade gold deposits in the Abitibi greenstone belt, Canada





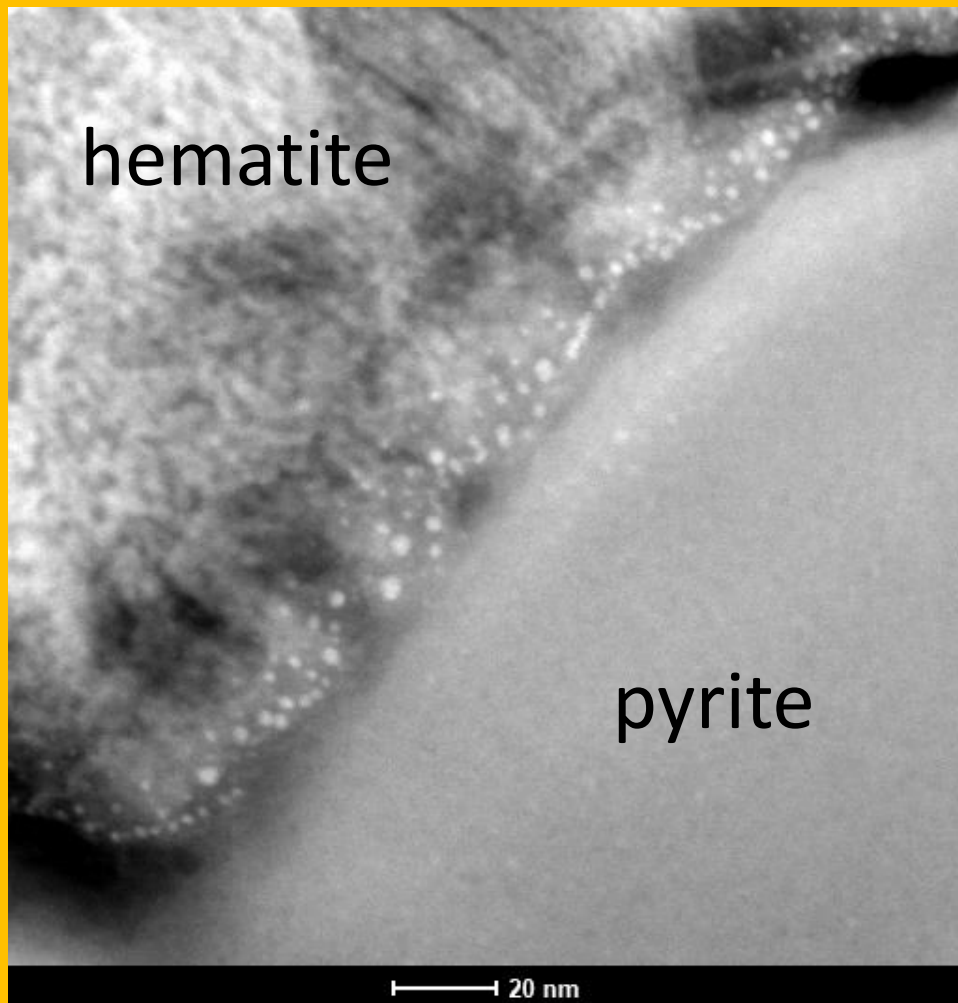
Gold-bearing Pyrite



Hastie et al. 2021
Communications Earth & Environment

Formation of nanoparticles along the pyrite-hematite interface



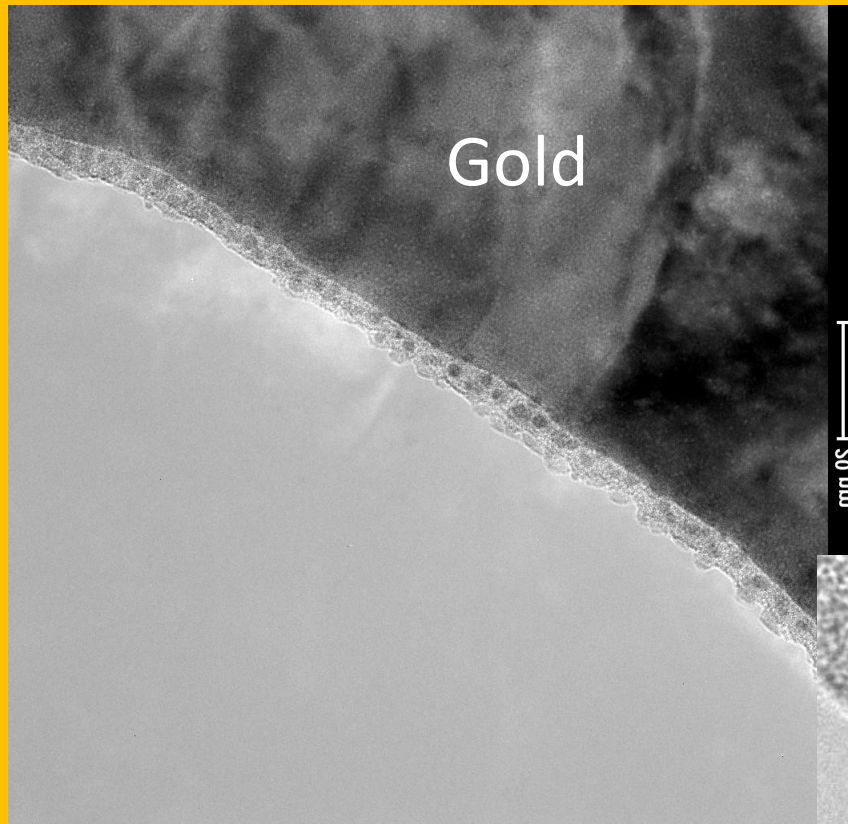


Gold nanoparticles only visible
in thinner part along interface

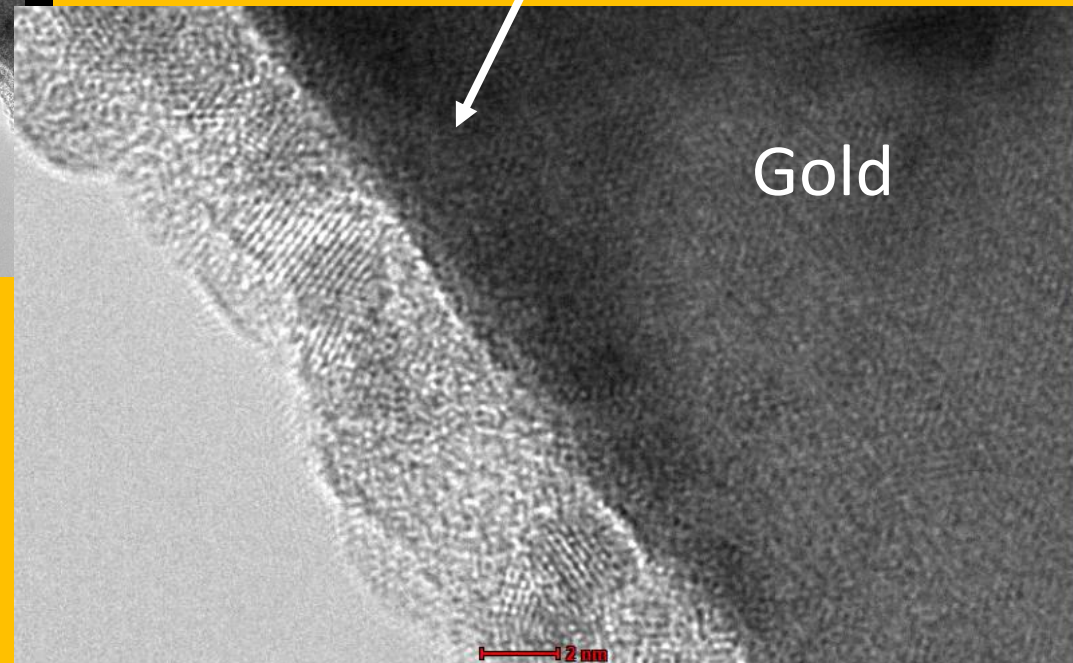


Nanoparticle size: 3-5 nm

Consequence of gold nanoparticle formation: gold coarsening through gold nanoparticle attachment

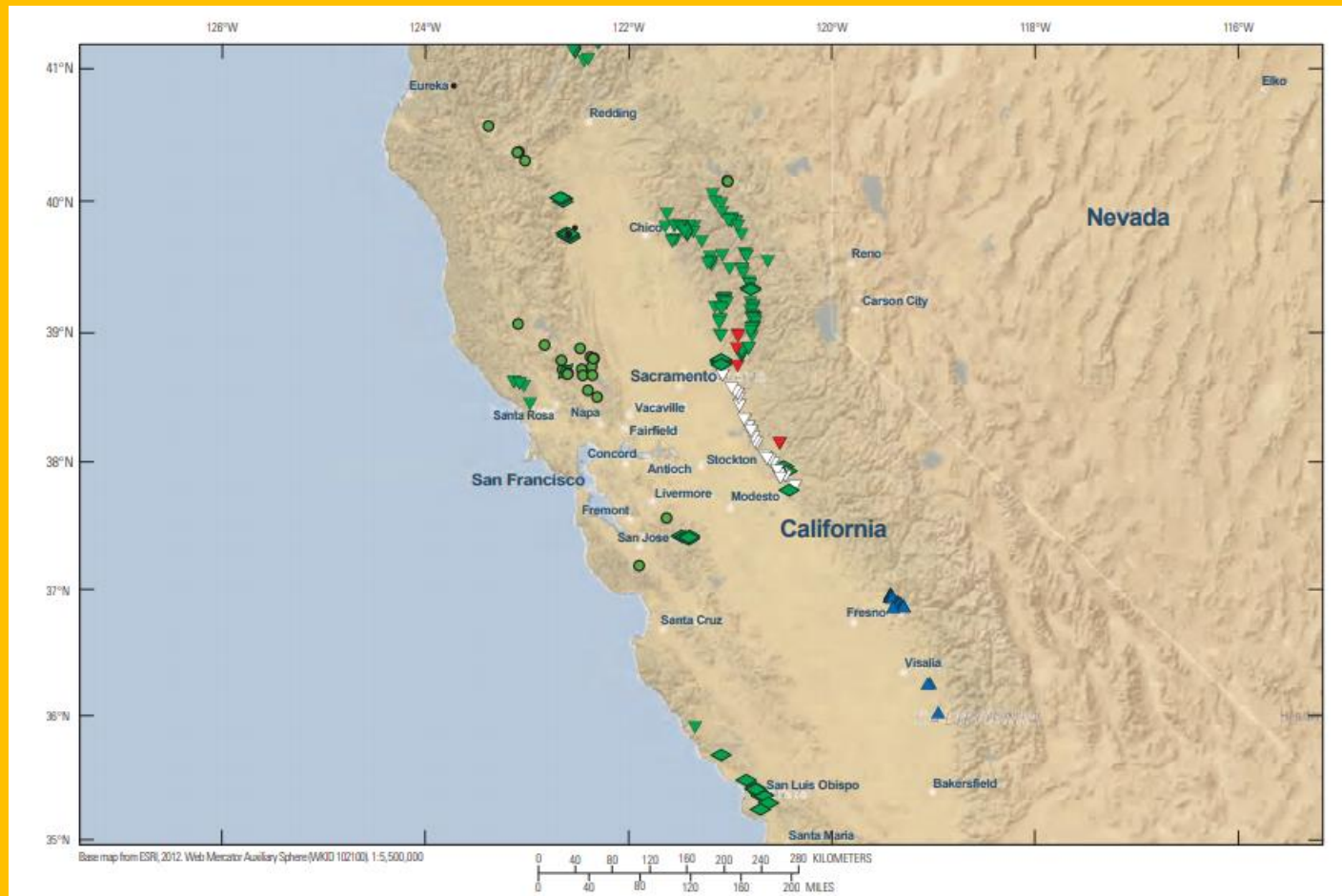


Gold nanoparticles
in FeOx layer

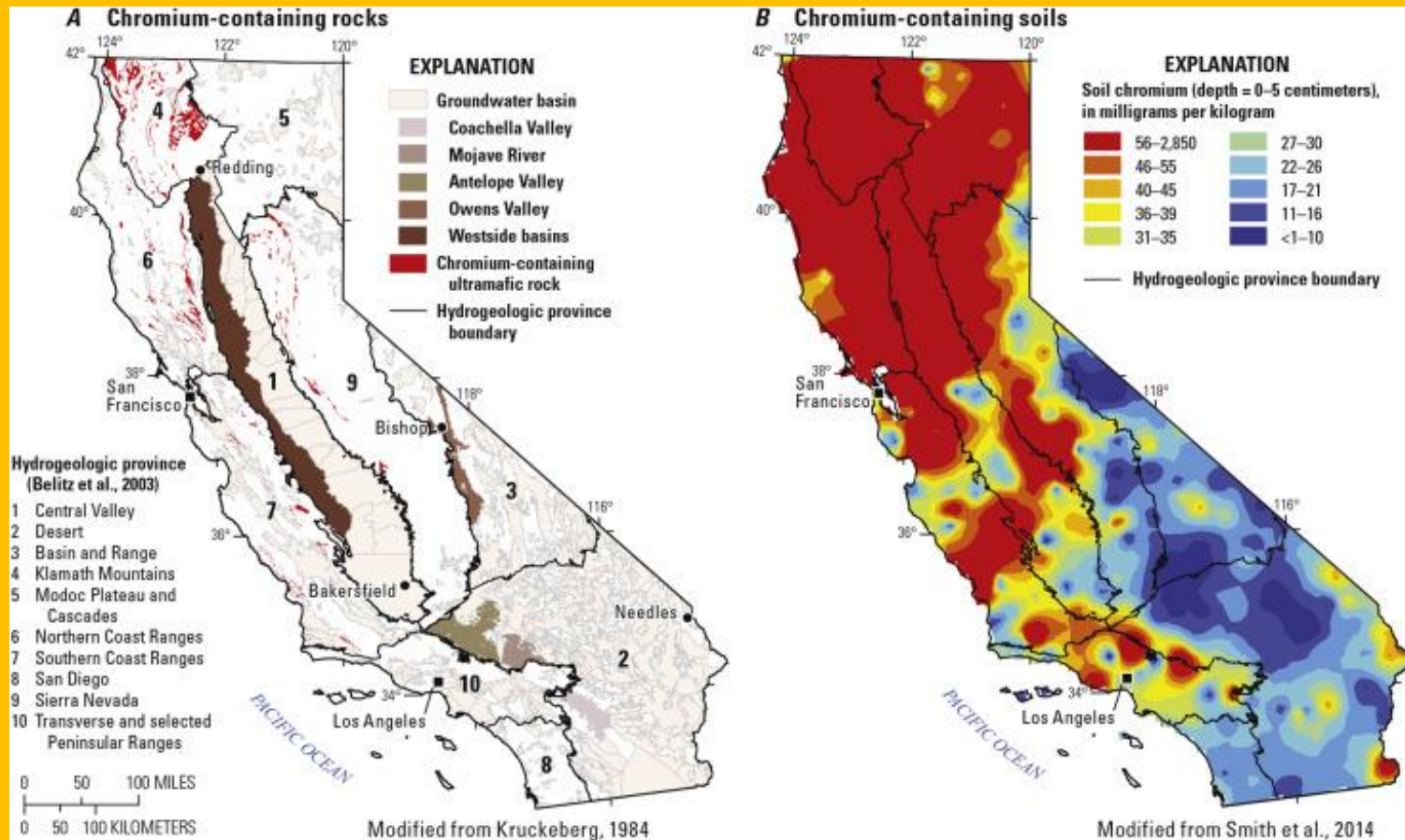


The formation of Chromite nanoparticles

Chromite ore deposits related to ultra mafic rocks
(ophiolites from the oceanic crust)



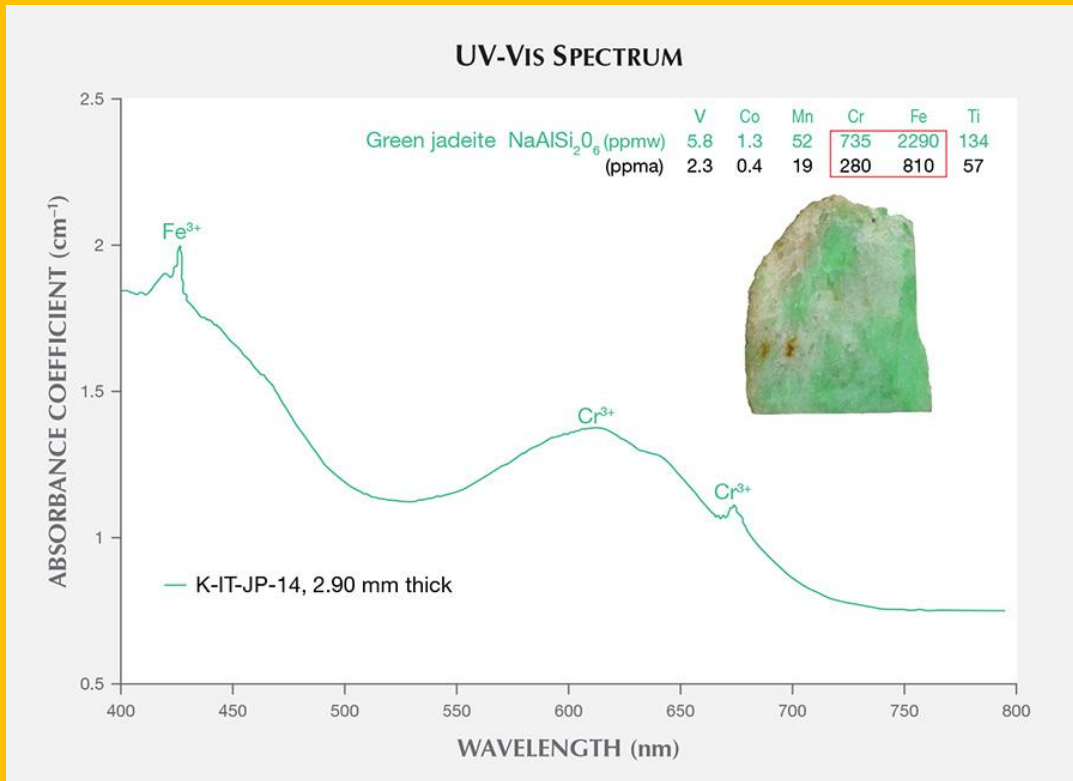
Cr concentrations in surficial soils in California



2850 mg/kg
maximum Cr

Izbicki et al. 2015, Applied Geochemistry 63, 203-217

Cr^{3+} structurally incorporated into pyroxenes, $A+B^{3+}\text{Si}_2\text{O}_6$



Cr^{3+} is structurally incorporated into pyroxenes

Jadeite
 $\text{NaAlSi}_2\text{O}_6$

(<https://www.gia.edu>)

Greenschist metamorphism

300–450 °C

2–10 kilobars

Mg-pyroxenes and -amphiboles

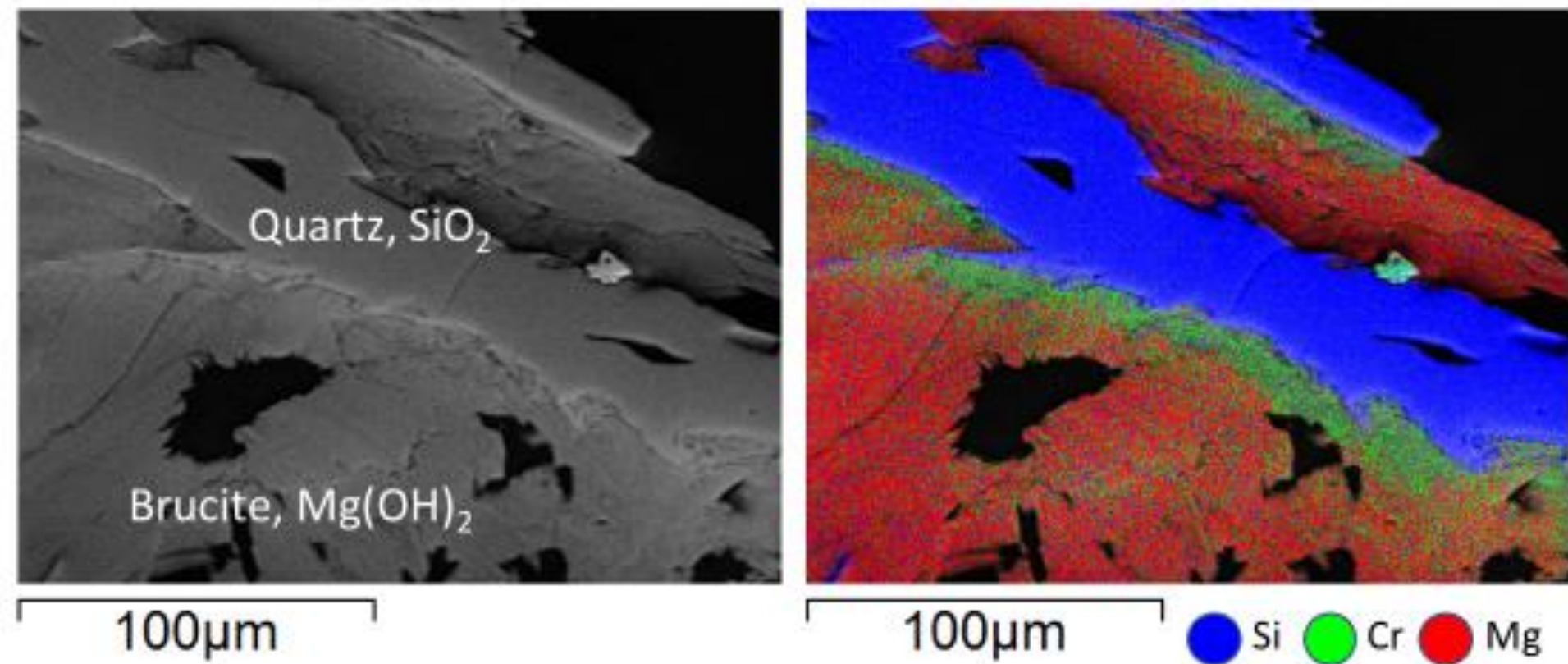


Mg-rich minerals of the chlorite- and
serpentine-group or
brucite $\text{Mg}(\text{OH})_2$ + quartz

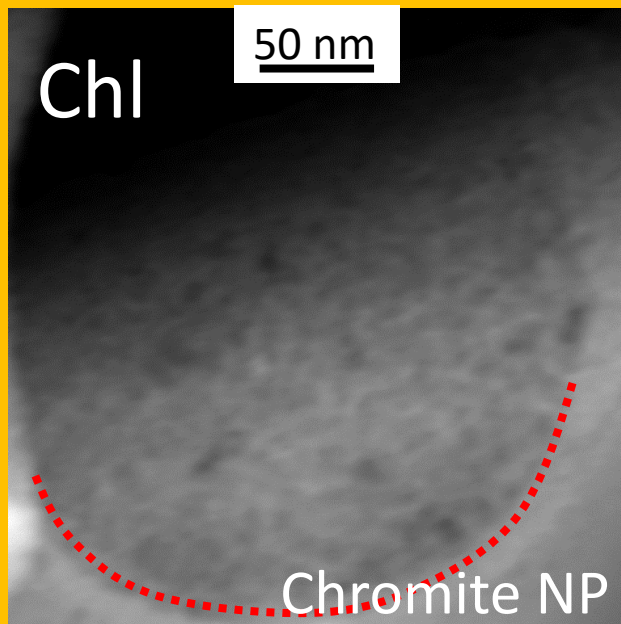
Transformation involves again dissolution-
reprecipitation and thus porosity



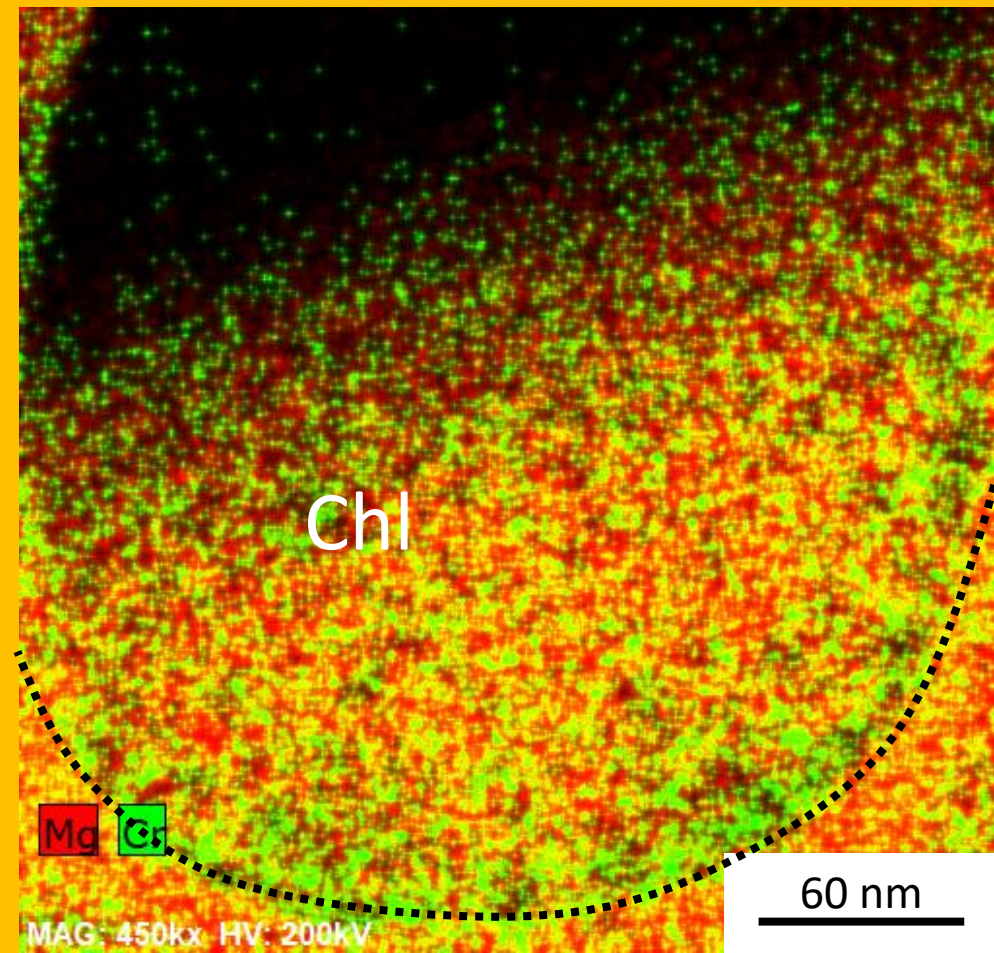
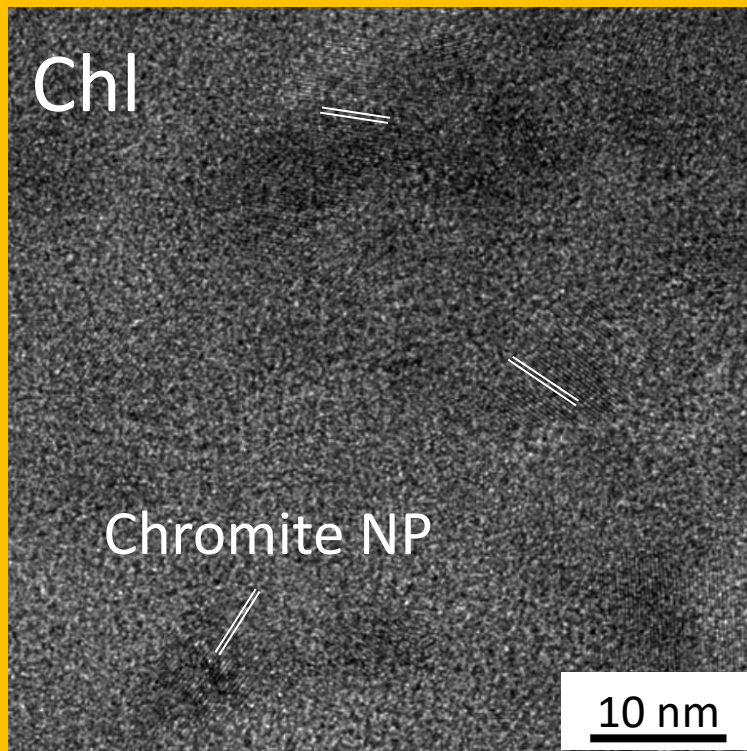
On the micrometer scale: enrichment of Chromium along phase boundaries



McClenaghan & Schindler 2021
American Mineralogist

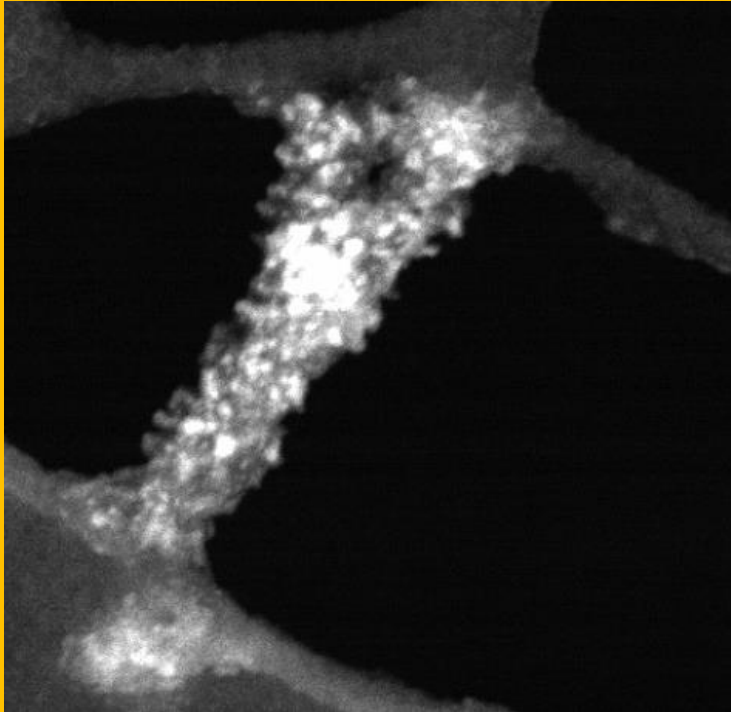


On the nanometer-scale:
chromite nanoparticles
along clinocllore grain
boundaries



Environmental Relevance

Release of chromite nanoparticles
during weathering of Cr-bearing clinochlore



Chromite nanoparticles attached
on amorphous Al-hydroxide colloid
after 3 months leaching of clinochlore
at pH =5

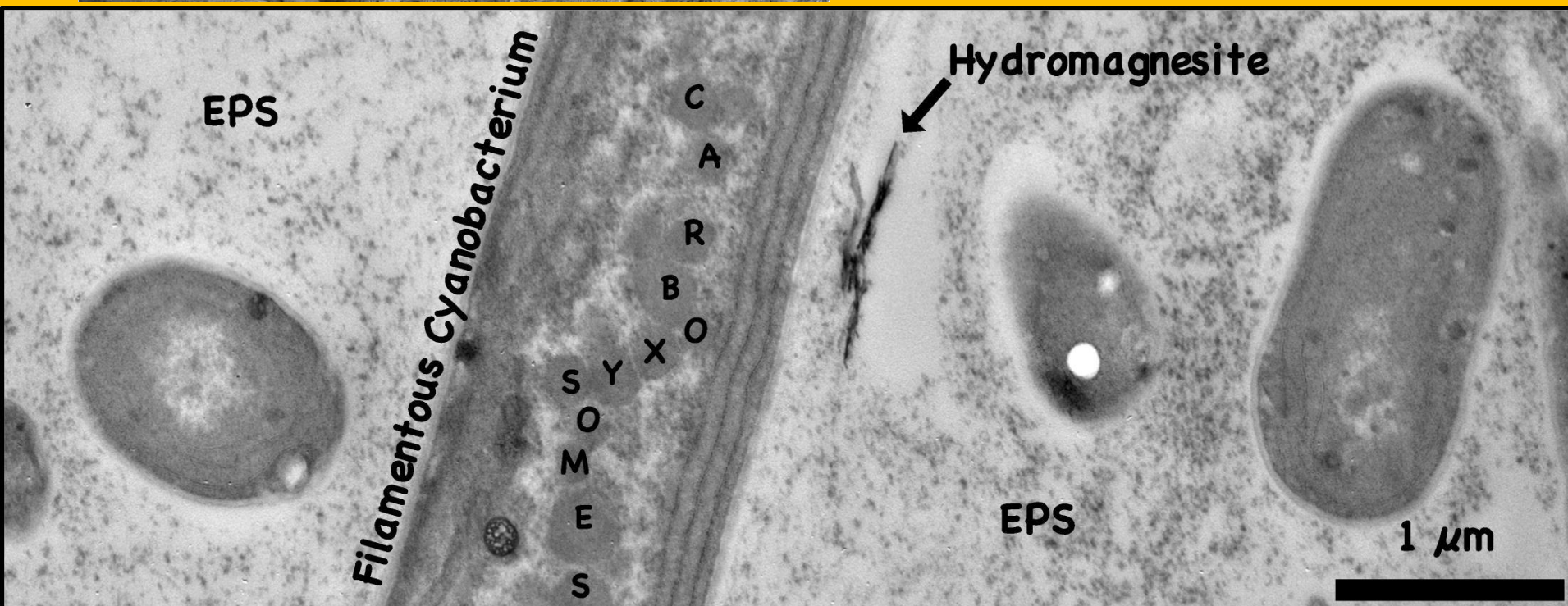
McClenaghan & Schindler 2021
American Mineralogist, in press

Nanoparticle formation
during low-T weathering
processes



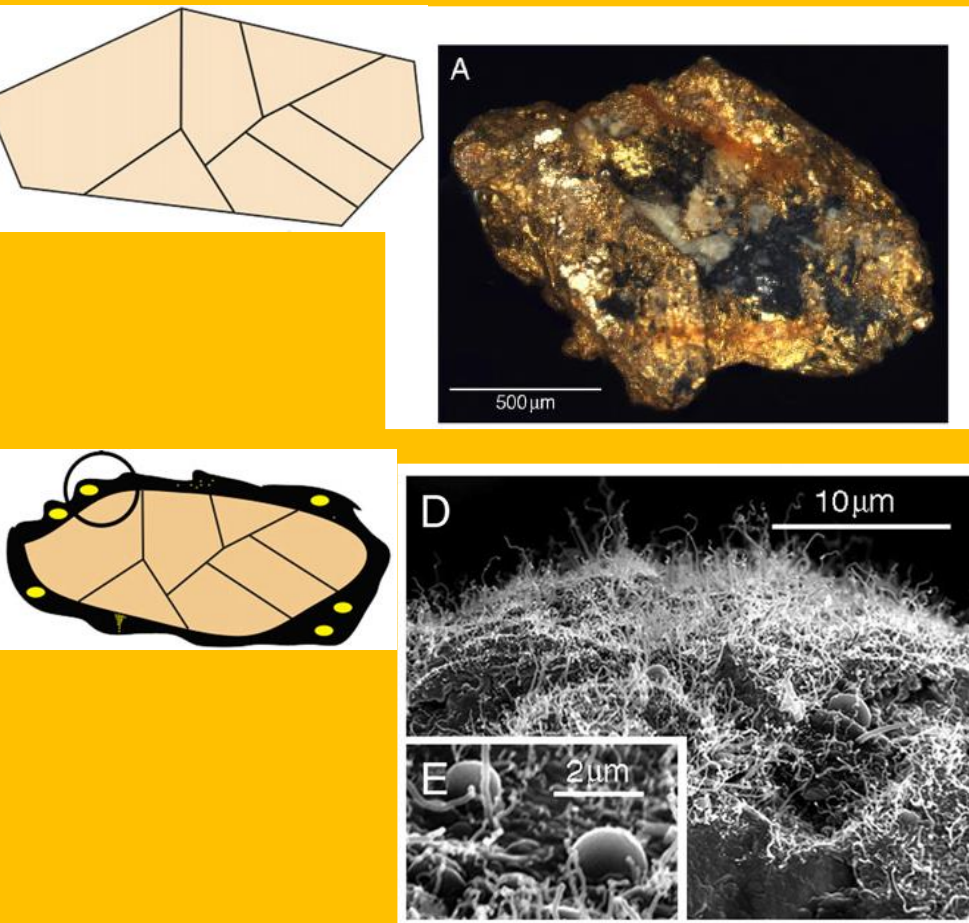
Interior and surroundings of a
bacterium (EPS)
may also contain
nano-size pore spaces

EPS = extracellular polymeric
substances



Low Temperature processes:

A. biotic-controlled weathering of gold in Australia



Primary gold with Ag and Cu
~80-90% Au



“Porosity” in Biofilms



Abiotic and biotic
processes in biofilms

“Purified” gold nanoparticles
Au > 95 wt% Au

Fairbrother et al. 2012
Chemical Geology 320–321 (2012) 17–31

Reith et al (2010) termed Biofilms:
“Nanoparticles factories”
Geology 38, 843–846.

Consequences

Cycles of Biofilm formation and destruction results in the release of gold as gold nanoparticles and thus in a high mobility of Au in many near surface environments

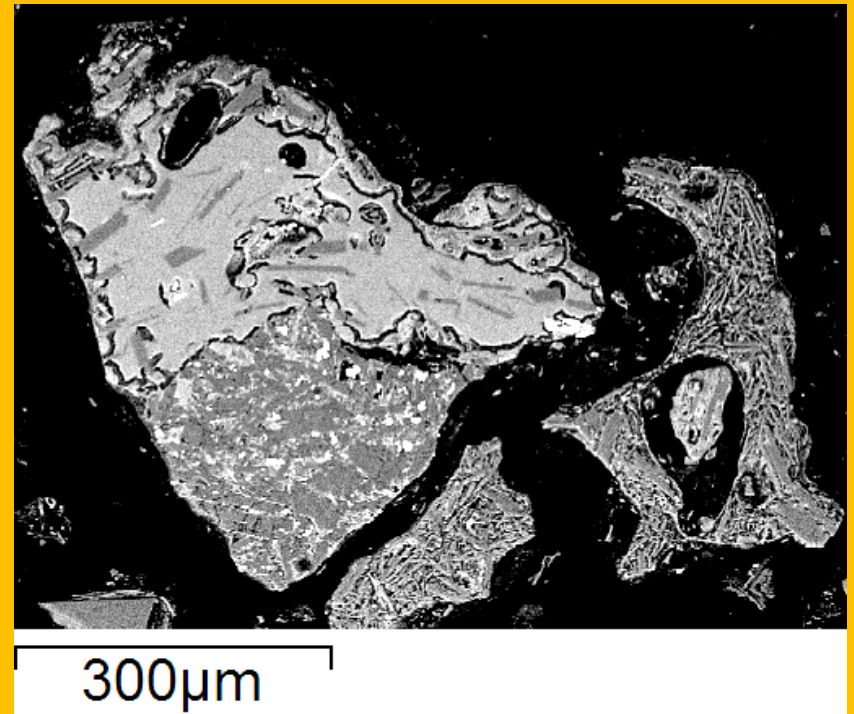
the formation of secondary Au deposits

the development of geochemical halos around buried mineralization

Low Temperature processes:

B. Abiotic-controlled weathering

Alteration of volcanic glass in a Martian analogue, JSC-1



Schindler et al. (2019) *Geochimica et Cosmochimica Acta* 264, 43–66

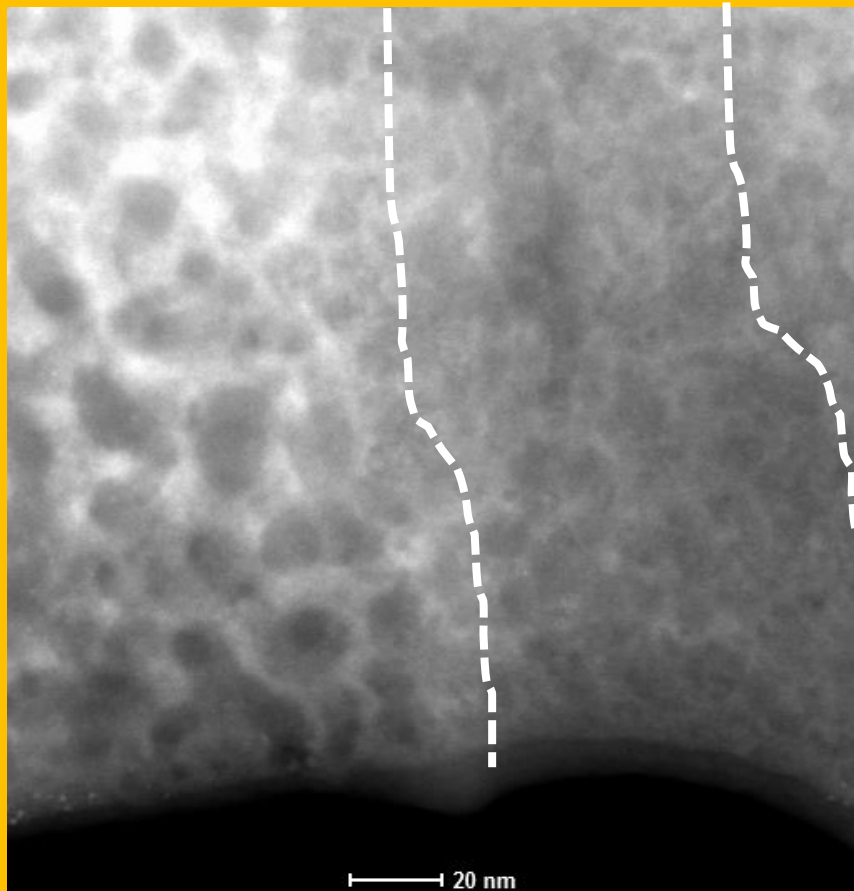
Dissolution of glass and re-precipitation of Ferrihydrite and hydrous amorphous silica

Porous
Alteration
layer

Reaction front

“glass”

← porosity

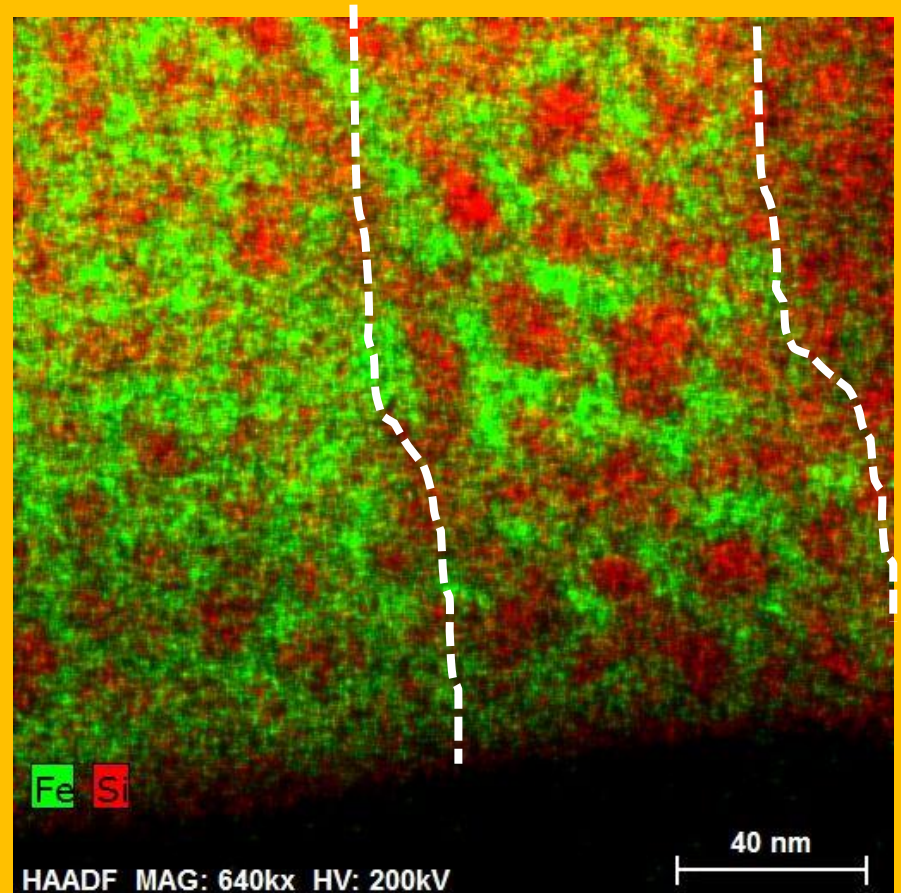


Reaction front

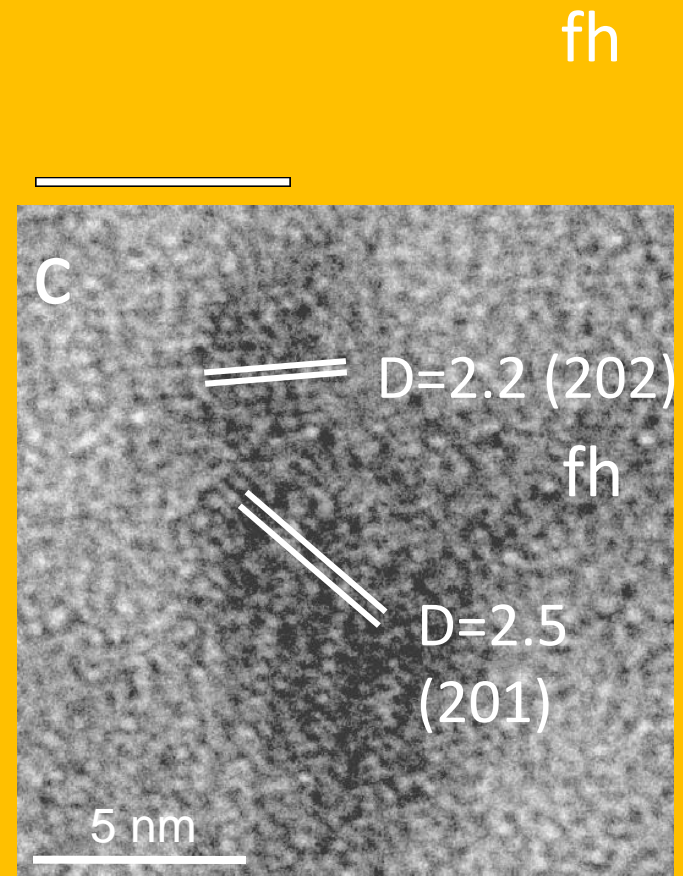
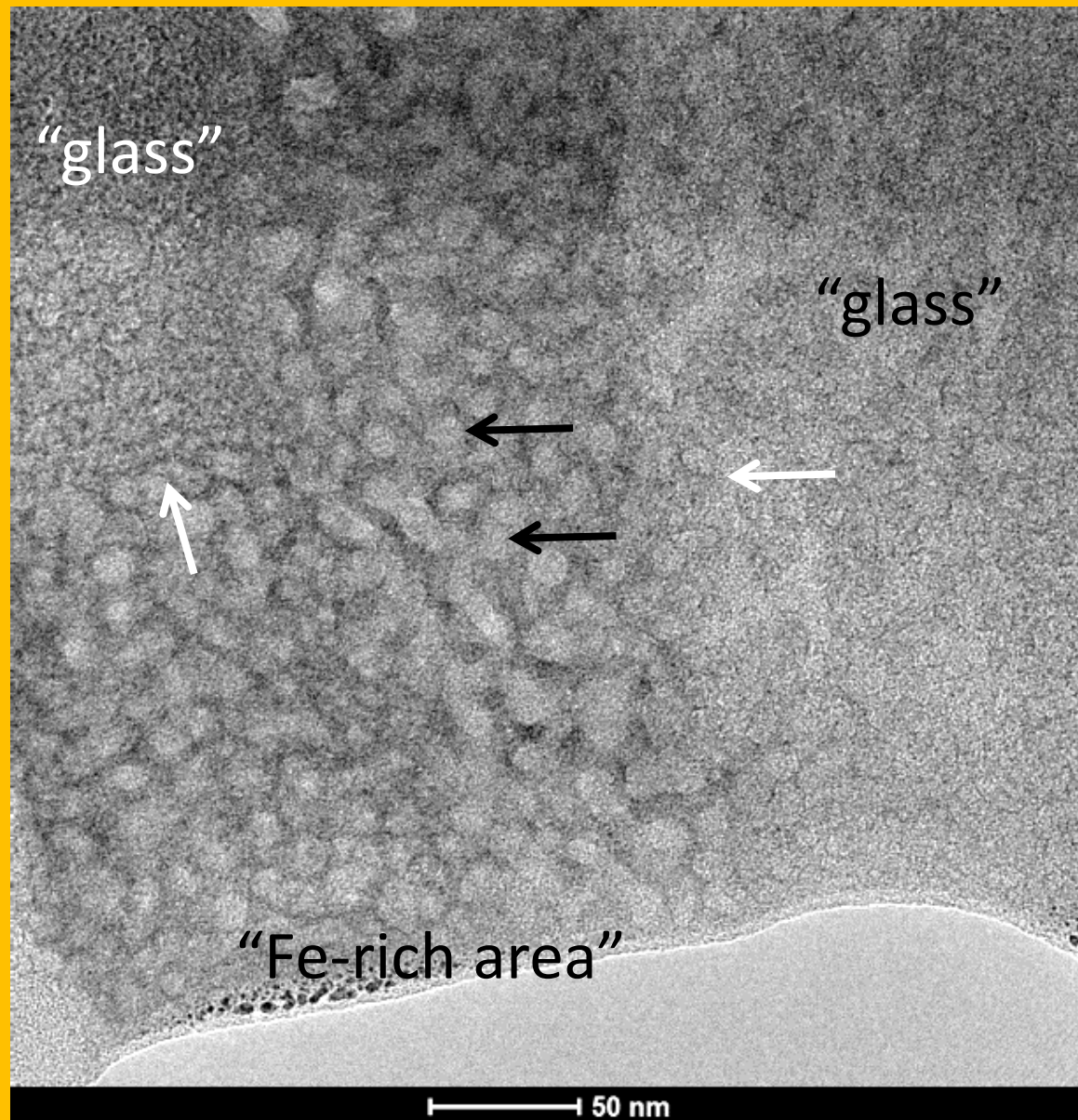
Fe-hydroxide
aggregation

Fe-hydroxide
formation

“glass”

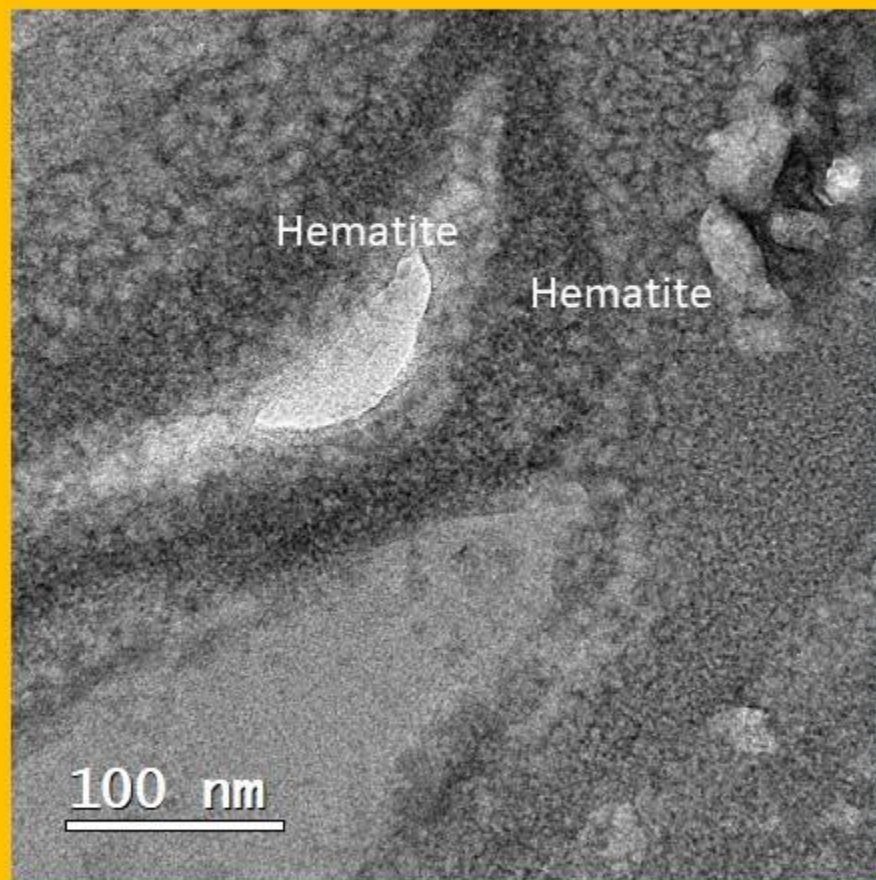


Ferrihydrite “nano-domains” in pore spaces between hydrous amorphous silica



Consequence:

Porosity created during the alteration of volcanic glass is one of the “birthplaces” of Fe-oxide nanoparticles in volcanic soils on Earth and perhaps on Mars



Schindler et al. *Geochimica et Cosmochimica Acta* 264, 43–66


Conclusions

One common feature:

Enrichment (sometimes termed supergene process) of an element in the form of nanoparticles as a consequent of re-equilibration

Au-bearing pyrite  Gold nanoparticles

Cr-bearing pyroxene  Chromite nanoparticles

Primary Au-Ag-Cu ore  High fineness gold nanoparticles

Fe-bearing Volcanic glass  Fe-oxide nanoparticles

...and many more examples can be found in the literature

Thank you to



NSERC
CRSNG

....and all my students and collaborators